

Units and Dimensional Analysis

- ❑ Math modeling is to used to solve real world problems. Most of quantities in the real world have units. Or physical quantities are measured using units.
- ❑ A **unit of measurement** is a definite magnitude of a physical quantity, defined and adopted by convention or by law, that is used as a standard for measurement of the same physical quantity. Any other value of the physical quantity can be expressed as a simple multiple of the unit of measurement.

Dimensional Analysis

□ Purposes:

- Check the correctness of mathematical models for physical problems by checking dimensional unit
- Derive mathematical models
- Reduce parameters through ***Non-dimensionalization process or (scaling)***
- Identify key parameters such as Reynolds number, peck number etc.

Units Examples

- ❑ [Length](#) is a physical quantity. The [meter](#) is a unit of length that represents a definite predetermined length. When we say 10 meters (or 10 m), we actually mean 10 times the definite pre-determined length called “meter (metre)”.
- ❑ The definition, agreement, and practical use of units of measurement have played a crucial role in human endeavour from early ages up to this day. Different systems of units used to be very common. Now there is a global standard, the [International System of Units](#) (SI), the modern form of the [metric system](#). But English system is still in use (UK, US, some counties in British Commonwealth Union)

Units

The recommended scientific system of units is the SI system, which includes 7 basic units.

Quantity	Unit	Symbol
Length	metre	m
Mass	kilogram	kg
Time	second	s
Electric current	ampere	A
Temperature	kelvin	K
Luminous intensity	candela	cd
Amount of substance	mole	mol

Large and small units

There is a wide range of sizes for all sorts of quantities. So we use special symbols for large and small multiples of the basic units.

Multiplication factor	Prefix	Symbol
10^{12}	tera	T
10^9	giga	G
10^6	mega	M
10^3	kilo	k
10^{-2}	centi	c
10^{-3}	milli	m
10^{-6}	micro	μ
10^{-9}	nano	n

Commonly used units

There are also other commonly used units which are combinations of some of the SI units.

Quantity	Unit	Symbol
Force	newton	N (kg m s^{-2})
Energy	joule	J ($\text{kg m}^2 \text{s}^{-2}$)
Power	watt	W (J s^{-1} or $\text{kg m}^2 \text{s}^{-3}$)
Frequency	hertz	Hz (s^{-1})
Pressure	pascal	Pa (N m^{-2} or $\text{kg m}^{-1} \text{s}^{-2}$)

Non-SI units

There are a number of non-SI units which are in common use by scientists and engineers.

Quantity	Unit	Symbol
Area	hectare	ha ($= 10^4 \text{ m}^2$)
Volume	litre	l ($= 10^{-3} \text{ m}^3$)
Volume	millilitre	ml ($= 10^{-6} \text{ m}^3$)
Temperature	degree Celsius	$^{\circ}\text{C}$ ($0^{\circ}\text{C} \approx 273 \text{ K}$)
Mass	gram	g ($= 10^{-3} \text{ kg}$)
Mass	tonne	t ($= 10^3 \text{ kg}$)
Energy	kilowatt hour	kW h ($= 3.6 \times 10^6 \text{ J}$)
Energy	electronvolt	eV ($\approx 1.6 \times 10^{-19} \text{ J}$)
Energy	calorie	cal ($= 4.1868 \text{ J}$)
Pressure	bar	bar ($= 10^5 \text{ Pa}$)
Pressure	atmosphere	atm ($\approx 1.013 \times 10^5 \text{ Pa}$)

Units Converting

The Fahrenheit scale of temperature

To convert from °F to °C,

$$^{\circ}\text{C} = 5(^{\circ}\text{F} - 32)/9 \sim (^{\circ}\text{F} - 32) / 2$$

To convert from °C to °F

$$^{\circ}\text{F} = 9^{\circ}\text{C} / 5 + 32 \sim 2^{\circ}\text{C} + 32$$

Converting any units (almost anything)

<http://www.onlineconversion.com/>

British units system

Another system of units which is still in use is the British system.

fps unit	SI equivalent
inch (in)	0.0254 m
foot (ft)	0.3048 m
mile (5280 ft) (5 miles \approx 8 km)	$1.609\,344 \times 10^3$ m
nautical mile (6080 ft)	$1.853\,184 \times 10^3$ m
acre	$4.046\,856 \times 10^3$ m ² (\approx 0.4 ha)
pint (pt)	$5.682\,613 \times 10^{-4}$ m ³
gallon (gal)	$4.546\,09 \times 10^{-3}$ m ³
ounce (oz)	$2.834\,952 \times 10^{-2}$ kg
pound (lb)	0.453 592 37 kg
horsepower (hp)	7.457×10^2 W

An example

The price of milk in the UK is about 1.65 pounds every 6 pints. That in China is 33 RMB every 6 litre. Assume that 1 pound = 15 RMB.

❑ Which is cheaper? (using the same unit)

2008: UK: $1.65/6 \rightarrow 15*1.65/3.409=7.26$, CH: $33/6=5.5$

2013: UK: $1.65/6 \rightarrow 9.37*1.65/3.409=4.53$, CH: $33/6=5.5$

❑ There is a 50% price rise in China recently.

Which is cheaper? CH: $33*1.5/6=5.5*1.5=8.25$

Dimensional Analysis

- ❑ A model which describes a physical, biological, economic or managerial system involves a variety of parameters or variables
- ❑ With each variable or parameter we can associate a dimension

$$\text{area} = \text{length}^2$$

$$\text{velocity} = \text{length} / \text{time}$$

Dimensions: Definition

- ❑ All ***mechanical*** quantities can be expressed in terms of the fundamental quantities
mass (M) or kg, length (L) or m, time (T) or s
- ❑ Other physical quantities can be expressed as a combination of these 3 terms
- ❑ The resultant combination is called the '*dimensions*' of that physical quantity

Dimensions: Definition

- We use square brackets [] to denote “the dimension of ”

$$[\text{area}] = L^2$$

$$[\text{speed}] = L T^{-1}$$

$$[\text{density}] = M L^{-3}$$

$$[\text{angle}] = L L^{-1} = L^0$$

$$[\text{force}] = M L T^{-2}$$

$$[\text{weight}] = M L T^{-2}$$

Note, dimensions are independent of the units used

Full Dimensional List

Mass	-	M
Length	-	L
Time	-	T
Electric Charge	-	Q
Temperature	-	θ
Number of Moles	-	MOL
Luminosity	-	?

Dimensional Analysis

- ❑ Any sensible equation must be dimensionally consistent

$$[\text{left-hand side}] = [\text{right-hand side}]$$

- ❑ It is a good idea to carry out this check on all the equations appearing in a model
- ❑ This reveals any modeling errors

Dimensional Analysis

- ❑ Addition of terms only makes sense if each term has the same dimensions
- ❑ For a proposed equation, each term must be checked for consistency

$$A = B + (C \times D)$$

- ❑ A, B and (C x D) must have the same dimensions

Determine the units for constants

Any constants appearing in equations can be

- Either be dimensionless (pure numbers)
- Or can have dimensions

Example

Suppose that we are modeling the force on a moving object due to air resistance. If we assume the magnitude of the force F is **proportional** to the **square** of the speed v :

$$F = kv^2$$

In dimensions: $[F] = [kv^2]$



$$MLT^{-2} = [k][LT^{-1}]^2 = [k]L^2T^{-2}$$

- * For consistency, we require $[k] = ML^{-1}$
- * k is measured in $kg\ m^{-1}$.

Dimensional Analysis

- ❑ If expressions involving $\exp(at)$ or $\sin(at)$ appear in our model, where t stands for time
- ❑ The parameter a must have dimensions T^{-1} so that at is a dimensionless number.
- ❑ If an equation involves a derivative, the dimensions of the derivative are given by the ratio of the dimensions
- ❑ If p is the pressure in a fluid at any point, z is the depth, then

Dimensional Analysis: example

$$\left[\frac{dp}{dz} \right] = \frac{[p]}{[z]} = \frac{\text{ML}^{-1}\text{T}^{-2}}{\text{L}} = \text{ML}^{-2}\text{T}^{-2}$$

$$\left[\frac{\partial p}{\partial t} \right] = \frac{[p]}{[t]} = \frac{\text{ML}^{-1}\text{T}^{-2}}{\text{T}} = \text{ML}^{-2}\text{T}^{-3}$$

$$\left[\frac{\partial^2 v}{\partial x^2} \right] = \frac{[v]}{[x]^2} = \frac{\text{L T}^{-1}}{\text{L}^2} = \text{L}^{-1}\text{T}^{-1}$$

Dimensional Analysis: Pendulum

Suppose that we are trying to develop a model which will predict the period t of a swinging pendulum:-

List of Factors (attention restricted to 4 factors)

- the length l
- the mass m
- the angle θ
- acceleration g due to gravity

- Assume that the period

$$[t] = [k l^a m^b g^c \theta^d]$$

where a, b, c, d and k are unknown real numbers

Dimensional Analysis: Pendulum

Considering dimensions, we have

$$[t] = [kl^a m^b g^c \theta^d]$$

θ is dimensionless and k is assumed to be as well, so

$$T = L^a M^b (LT^{-2})^c$$

Equating powers of M, L and T on both sides

$$\left. \begin{array}{l} L : 0 = a + c \\ M : 0 = b \\ T : 1 = -2c \end{array} \right\} \Rightarrow b = 0, \quad c = -\frac{1}{2}, \quad a = -c = \frac{1}{2}$$

Dimensional Analysis: Pendulum

$$[t] = [kl^a m^b g^c \theta^d]$$

therefore

$$t = kl^{1/2} g^{-1/2} \theta^d$$

At present, d is unresolved, i.e., it can take any value

Summing terms of this form leads to the general result

$$t = f(\theta) l^{1/2} g^{-1/2}$$

Dimensional Analysis: Fluid

Example: The pressure p at a depth h below the surface of a fluid of density ρ is given by $p = \rho gh$, where g is the acceleration due to gravity. Check the dimensions.

$$[p] = \left[\frac{\text{force}}{\text{area}} \right] = \frac{\text{MLT}^{-2}}{\text{L}^2} = \text{ML}^{-1}\text{T}^{-2}$$

$$[\rho] = \text{ML}^{-3}$$

$$[g] = [\text{acceleration}] = \text{LT}^{-2}$$

$$[h] = \text{L}$$

$$[\rho gh] = \text{ML}^{-3}\text{LT}^{-2}\text{L} = \text{ML}^{-1}\text{T}^{-2}$$

The dimensions are consistent.

Nondimensionalization

- ❑ **Nondimensionalization** is the partial or full removal of [units](#) from an [equation](#) involving [physical quantities](#) by a suitable substitution of [variables](#). This technique can simplify and [parameterize](#) problems where [measured](#) units are involved. It is closely related to [dimensional analysis](#). In some physical [systems](#), the term **scaling** is used interchangeably with *nondimensionalization*, in order to suggest that certain quantities are better measured relative to some appropriate unit. These units refer to quantities [intrinsic](#) to the system, rather than units such as [SI](#) units.
- ❑ Nondimensionalization can also recover characteristic properties of a system. The technique is especially useful for systems that can be described by [differential equations](#).
- ❑ Nondimensionalization can reduce the number of parameters and keep the most important ones such as Reynolds number.

Nondimensionalization steps

To nondimensionalize a system of equations, one must do the following:

1. Identify all the independent and dependent variables;
2. Replace each of them with a quantity scaled relative to a characteristic unit of measure to be determined;
3. Divide through by the coefficient of the highest order polynomial or derivative term;
4. Choose judiciously the definition of the characteristic unit for each variable so that the coefficients of as many terms as possible become 1;
5. Rewrite the system of equations in terms of their new dimensionless quantities.

An illustrative example

$$a \frac{dx}{dt} + bx = Af(t).$$

1. In this equation the independent variable here is t , and the dependent variable is x .

2. Set $x = \chi x_c$, $t = \tau t_c$. This results in the equation

$$a \frac{x_c}{t_c} \frac{d\chi}{d\tau} + bx_c \chi = Af(\tau t_c) \stackrel{\text{def}}{=} AF(\tau).$$

3. The coefficient of the highest ordered term is in front of the first derivative term. Dividing by this gives

$$\frac{d\chi}{d\tau} + \frac{bt_c}{a} \chi = \frac{At_c}{ax_c} F(\tau).$$

4. The coefficient in front of χ only contains one characteristic variable t_c , hence it is easiest to choose to set this to unity first:

$$\frac{bt_c}{a} = 1 \Rightarrow t_c = \frac{a}{b}. \text{ Subsequently, } \frac{At_c}{ax_c} = \frac{A}{bx_c} = 1 \Rightarrow x_c = \frac{A}{b}.$$

5. The final dimensionless equation in this case becomes completely independent of any parameters with units:

$$\frac{d\chi}{d\tau} + \chi = F(\tau).$$

Dimensional Analysis

- ❑ Well known to any engineer
- ❑ This simple information is also remarkably useful in mathematical modelling
- ❑ Will consider ideas in context of the modelling of physical systems, but are easily extended to any application

Dimensional Analysis

- ❑ Check the validity of equations proposed during the modelling process
- ❑ Find a number of independent parameter groups (and calculate them)
- ❑ Find the relative sizes of various terms when model equations are proposed
- ❑ Normalise problems in terms of non-dimensional variables whose typical scale is of the order of one, and hence simplify them